STEREOCHEMISTRY OF THE C3 ADDITION OF 1-LITHIO 1,1-BIS METHYLSELENOALKANES

TO METHYL CYCLOHEXENONES

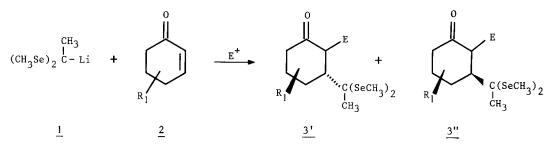
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The presence of DME or HMPT allows the C_3 regio and stereoselective introduction of 1,1-bis methylselenoalkanes to methyl cyclohexenones. The resulting compounds have been stereoselectively transformed to 3-ethyl methyl cyclohexanones.

Selective introduction of organometallics at the C₃ site of enones is of great synthetic value. This has been achieved by addition of copper salts ¹ for organometallics bearing an hydrocarbon residue and more recently, for α -heterosubstituted organolithiums ² which do not bear an extra stabilizing group, by use of solvent which strongly solvates the metal (e.i. HMPT and DME in some cases).

Nothing is known at the present time about the stereochemistry of the addition of the last species to alkylsubstituted enones. We report here our preliminary observations concerning the stereochemistry of the addition of 1,1-bis (methylseleno) 1-lithio ethane <u>1</u> (prepared from trismethylseleno ethane and n-BuLi in THF or DME at -78° C)³ to methyl cyclohexenones <u>2</u> [respectively in THF/HMPT (1.1 eq) (method A) or in DME (method B) at -78° C and quenching with aqueous NH₄Cl at -78° C] ^{3b}.



Except for the 3-methyl cyclohexenone, the C_3 adducts are isolated in 60 to 76% yield (Table) and little if any allylic alcohols resulting from the competing C_1 attack is formed when method A is used. A good to high stereoselectivity (ranging from 68-32 to 99-1) is observed (¹H, ¹³C-NMR, $|CC|^2$ OV1) depending upon the nature of the starting enone (Table). The experimental conditions [e.i. solvent change THF/HMPT or DME), temperature used (-110°C,

-78°C, -45°C or -78°C to 20°C) and the time of the reaction (3-5 seconds or 10 min.)] do not seem to affect the ratio of isomers present after quenching.

We have also prepared the δ -ketoacetals <u>3a</u> by reaction of <u>1</u> on cyclohexenone in THF/HMPT followed by the trapping of the resulting lithic enclate ⁴ with methyl iodide (5 eq) in THF containing HMPT (5 eq, 0°, 1hr)⁵. The product consists quite exclusively of one single stereoisomer <u>3a'</u> which is the minor isomer (1%) obtained from the reaction of the same organometallic 1 on 2-methyl cyclohexenone (Table, entries 1 and 2).

The trans isomer 3' is the major one in case of 4- and 5-methyl-cyclohexenones (Table, entries 4, 5) and on methylation of the enolate formed by reaction of 1 with cyclohexenone (Table, entry 1).

The cis isomer 3'' being mainly formed in the case of 2- and 6-methylcyclohexenones (Table, entries 2, 6).

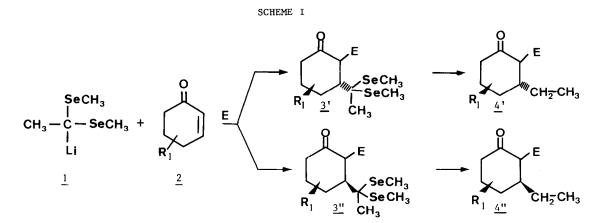
The stereochemical assignments have been performed on 3-ethyl-methyl-cyclohexanones $\underline{4}$ regio and stereoselectively prepared by tributyltin hydride reduction of the selenoacetal moiety of the δ -keto selenoacetals $\underline{3}$ [Bu₃SnH, 3 eq, AIBN (0.12 eq), toluene, 80°C, 0.3 hr]^{3,6} (see Table).

These assignments have already been done by Rivière ^{7,11} in the case of 3-ethyl-4-methyl cyclohexanone <u>4b</u> and we have correlated (see scheme II) the 3-ethyl-5-methyl-cyclohexanone <u>4c</u> with the 3,5-dimethyl cyclohexanone <u>7</u>, the structure of which was already determined by House ⁸. Thus, the enolate <u>5</u> prepared from 5-methyl cyclohexenone and bis (methylseleno)methyllithium (THF/HMPT, 1.1 eq, -78°C) was simultaneously transformed (scheme II) to the previously known 3,5-dimethylcyclohexanone <u>7</u> [by hydrolysis to <u>6</u> and selective reduction of the selenoacetal moiety] and to 3-ethyl-5-methyl-cyclohexanone <u>4c</u> [by trapping with chlorotrimethylsilane, cleavage of the selenoacetal moiety, selective methylation of the resulting α -selenoalkyllithium ⁹, further hydrolysis of the silylenol ether and reduction of the selenyl moiety].

The isomeric ratios of 7 and 4c are identical 7'/7'' = 4c'/4c'': 92/8) and the derivatives 4c of different origin are identical.

Finally 3-ethyl-2-methyl cyclohexanone $\underline{4a}$ and 3-ethyl 6-methyl cyclohexanone $\underline{4d}$ of various origins have been subjected to epimerization reactions (H_2SO_4 /acetone/60°, 0.05 hr) and we have assigned the trans stereochemistry to the major isomer : pure $\underline{4a'}$ and $\underline{4a''}$ (Table, entries 1 and 2) produced a 77/23 ratio of $\underline{4a'}/\underline{4a''}$ while pure $\underline{4d'}$ and $\underline{4d''}$ (Table, entry 6) gave a 68/32 ratio of $\underline{4d'}/\underline{4d''}$.

The reactions described allow the highly stereoselective introduction of groups bearing a latent fonctionality at the C_3 site of an enone 3b . Moreover, the cleavage of the carbonselenium bond in <u>8</u> (scheme II) is of particular interest since it is the first reported cleavage of a functionalized selenoacetal. Work is now in progress to use the potentiality of the reported reactions for the synthesis of natural products.

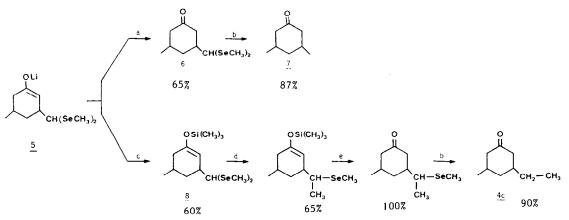


ΤA	В	L	E

Entry	R ₁	E	Yields in <u>3</u> (Method)	Trans/cis Ratio Method A or B	Yields in <u>4</u>	Trans/cis Ratio
1	Н	CH3	70 (A) [*]	<u>3a'/3a"</u> : 99/1	95	4a'/4a" : 99/1
2	2-СН ₃	Н	65 (A),51 (B)	<u>3a'/3a"</u> : 1/99	87	4a'/4a" : 1/99
3	3-сн ₃	Н	0 (A,B)	-	-	-
4	4-сн ₃	Н	58 (A),47 (B)	<u>3b'/3b"</u> : 95/5	74	4b'/4b" : 95/5
5	5-CH3 `	Н	76 (A),64 (B)	3c'/3c" : 87/13	90	4c'/4c" : 88/12
6	6-сн ₃	Н	74 (A),67 (B)	3d'/3d" : 32/68	85	4d'/4d" : 32/68

* Then CH₃I-HMPT (5eq,5eq) in THF, 0°, lhr. ** Each isomer can be obtained pure by preparative layer chromatography (SiO₂, ether/pentane 3/7 rf 0.62/0.57:<u>3d'/3d"</u>) and have been reduced stereoselectively to the corresponding <u>4d</u>.

SCHEME II



a. NH4Cl/H2O - b. Bu3SnH,AIBN,Toluene,80°.0.2hr - c. (CH3)3SiCl/Etg4 - J. n-duLi leq.then CH3I excess - e. 5% pTSOH/CH3OH/H2O.

References and notes

- Conjugate Addition Reactions of Organocopper Reagents, G.H. Posner, Organic Reaction, <u>19</u>, 1. 1972, J. Wiley & Sons, N.Y., ISBN 0-471-19619-3.
- Synthetic Methods Using α-Heterosubstituted Organometallics, A. Krief, Tet. Report, <u>94</u>, Tetrahedron, 36, 2531 (1980).
- a) D. Van Ende, A. Cravador and A. Krief, J. Organomet. Chem., <u>177</u>, 1 (1979) and references cited.
 - b) J. Lucchetti, W. Dumont and A. Krief, Tet. Lett., 2695 (1979).
- Ketone enolates: Regiospecific preparation and synthetic uses, J. d'Angelo, Tet. Report, 25, Tetrahedron <u>32</u>, 2979 (1976).
- 3-ethyl 2-methyl cyclohexanone 3a' is obtained in 25% yield when the reaction is performed with ! eq instead of 5 eq of HMPT.
- For reduction of selenoacetals to alkanes by triphenyltin hydride/AIBN, D.L.J. Clive, G.J. Chittattu, V. Farina, W.A. Kiel, S. Menchen, C.G. Russell, A. Singh, C.K. Wong and N.J. Curtio, J. Amer. Chem. Soc., <u>102</u>, 4438 (1980).
- 7. H. Rivière and J. Tostin, Bull. Soc. Chim. France, 568 (1969).
- 8. H.O. House and W.F. Fisher, J. Org. Chem., 33, 949 (1968).
- 9. M. Sevrin, D. Van Ende and A. Krief, Tet. Lett., 2643 (1976).
- 10. W. Cocker, T.B.H. Mc Murry and E.R. Simmons, J. Chem. Soc., 3022 (1965).
- 11. We have also prepared 3-ethyl cyclohexanones 4 from various methylcyclohexenones 2 and ethyl magnesium bromide in ether containing catalytic amount of CuCl. This work will appear in the full paper.

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